Water Report Input

Hover cursor over cells w/ red triangles to display helpful comments

Cations	Enter Ion Concentrations from Water Report (mg/L or ppm)		Anions		
Calcium (Ca)	39.0	146.3	Bicarbonate (HCO ₃)		
Magnesium (Mg)	12.0	0.7	Carbonate (CO ₃)		
Sodium (Na)	21.0	18.0	Sulfate (SO ₄)		
Potassium (K)	2.0	51.0	Chloride (CI)		
Iron (Fe)	0.0	0.9	Nitrate (NO ₃)		
		0.0	Nitrite (NO ₂)		
		0.0	Fluoride (F)		
If water report provides only Total Alkalinity (as CaCO ₃), use the calculator below to estimate the Bicarbonate and Carbonate concentrations. Insert the estimated results in the table above.					
Reported Total Alkalinity (as CaCO3) (mg/L or	Reported or Measured Water	Estimated Bicarbonate	Estimated Carbonate Concentration (ppm)		
ppm)	рН	Concentration (ppm)	Estimated Carbonate Concentration (ppm)		
121.0	8.0	146.3	0.7		

Ion Balance Results				
Total Cations (meq/L)	3.90	0.35	Cation/Anion Difference	
Total Anions (meq/L)	4.25			

Hardness and Alkalinity Results					
Total Hardness, as CaCO ₃ , (ppm)	147	122	Alkalinity (ppm as CaCO ₃)		
Permanent Hardness, as CaCO ₃ , (ppm)	25	122	RA Effective Hardness, (ppm as CaCO ₃)		
Temporary Hardness, as CaCO ₃ , (ppm)	122	87	Residual Alkalinity (RA), (ppm as CaCO ₃)		

Ion Concentration Conversion Calculator

Input Reporting Unit	Input	Output	Output Reporting Unit
Calcium (ppm as CaCO ₃)	39.0	15.6	Calcium (ppm)
Magnesium (ppm as CaCO ₃)	12.0	2.9	Magnesium (ppm)
Bicarbonate (ppm as CaCO ₃)	146.3	178.5	Bicarbonate (ppm)
Carbonate (ppm as CaCO ₃)	0.7	0.4	Carbonate (ppm)
Sulfate (ppm as SO₄⁻S)	6.0	18.0	Sulfate (ppm)
Nitrate (ppm as NO₃N)	0.9	4.0	Nitrate (ppm)
German Hardness (GH) (degrees)	0.0	0.0	Calcium (ppm)
Karbonate Hardness (KH) (degrees)	0.0	0.0	Bicarbonate (ppm)
Hardness (meq/L) or (mval)	0.0	0.0	Calcium (ppm)
Alkalinity (meq/L) or (mval)	121.0	7381.0	Bicarbonate (ppm)

Copyright © 2013 Martin Brungard . All rights reserved. Reproduction in whole or in part in any form or medium without the express written permission of Martin Brungard is prohibited.