



## Gas Permeability Testing – Flexelene™ 135C Material

Testing provided by:

Akron Rubber Development Laboratory  
2887 Gilchrist Rd.  
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ISO 9001:2008 registered  
Accredited Laboratory  
Member ACIL

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### **GAS PERMEABILITY-ASTM D1434-82 (2003), PROCEDURE V**

Apparatus: Custom Scientific Model CS-135

Gas Used: **CARBON DIOXIDE**

Test Temperature: 23.0°C

Test Gas Pressure: 30.0 psi

Permeation area of the sample: 66.4 cm<sup>2</sup>

Capillary Diameter: 0.0932 cm

Gas permeability testing was performed following the test conditions listed above. Please see the attached page explaining the calculations for the permeability constant.

#### **SAMPLE THICKNESS (CM) P (CM<sup>2</sup>/SEC-ATM) P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]**

<b>SAMPLE</b>	<b>THICKNESS (CM)</b>	<b>P (CM<sup>2</sup>/SEC-ATM)</b>	<b>P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]</b>
135C	0.1906	1.51877E-07	1.49891E-12

### **GAS PERMEABILITY-ASTM D1434-82 (2003), PROCEDURE V**

Apparatus: Custom Scientific Model CS-135

Gas Used: **METHANE**

Test Temperature: 23.0°C

Test Gas Pressure: 30.0 psi

Permeation area of the sample: 66.4 cm<sup>2</sup>

Capillary Diameter: 0.0932 cm

Gas permeability testing was performed following the test conditions listed above. Please see the attached page explaining the calculations for the permeability constant.

#### **SAMPLE THICKNESS (CM) P (CM<sup>2</sup>/SEC-ATM) P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]**

<b>SAMPLE</b>	<b>THICKNESS (CM)</b>	<b>P (CM<sup>2</sup>/SEC-ATM)</b>	<b>P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]</b>
135C	0.1906	1.18661 E-07	1.17109E-12

**GAS PERMEABILITY-ASTM D1434-82 (2003), PROCEDURE V**

Apparatus: Custom Scientific Model CS-135

Gas Used: **NITROGEN**

Test Temperature: 23.0°C

Test Gas Pressure: 30.0 psi

Permeation area of the sample: 66.4 cm<sup>2</sup>

Capillary Diameter: 0.0932 cm

Gas permeability testing was performed following the test conditions listed above. Please see the attached page explaining the calculations for the permeability constant.

**SAMPLE THICKNESS (CM) P (CM<sup>2</sup>/SEC-ATM) P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]**

SAMPLE	THICKNESS (CM)	P (CM <sup>2</sup> /SEC-ATM)	P [(CM <sup>3</sup> )(CM)/(CM <sup>2</sup> )(SEC)(PA)]
135C	0.1906	6.21960E-08	6.13827E-13

**GAS PERMEABILITY-ASTM D1434-82 (2003), PROCEDURE V**

Apparatus: Custom Scientific Model CS-135

Gas Used: **OXYGEN**

Test Temperature: 23.0°C

Test Gas Pressure: 30.0 psi

Permeation area of the sample: 66.4 cm<sup>2</sup>

Capillary Diameter: 0.0932 cm

Gas permeability testing was performed following the test conditions listed above. Please see the attached page explaining the calculations for the permeability constant.

**SAMPLE THICKNESS (CM) P (CM<sup>2</sup>/SEC-ATM) P [(CM<sup>3</sup>)(CM)/(CM<sup>2</sup>)(SEC)(PA)]**

SAMPLE	THICKNESS (CM)	P (CM <sup>2</sup> /SEC-ATM)	P [(CM <sup>3</sup> )(CM)/(CM <sup>2</sup> )(SEC)(PA)]
135C	0.1906	1.23786E-07	1.22167E-12

# **GAS PERMEABILITY-ASTM 01434-82 (2003), PROCEDURE V**

This is an example showing how the permeability constant has been calculated using a hypothetical slope. The permeability constant is calculated at steady state by the following steps.

1) The slope of the line from a graph plot of em Hg versus time is acquired. The volume-flow rate, q, is obtained by multiplying the slope by the cross-sectional area of the capillary tube.

For example:

$$q = 1.39 \times 10^{-5} \text{ em/sec.} \times (0.05 \text{ cm})^2 \times 3.14 = 1.09 \times 10^{-7} \text{ cm}^3/\text{sec.}$$

2) To convert to STP conditions, a barometric pressure of one atmosphere is used. For example, at 26 C:

$$q \text{ (STP)} = 1.09 \times 10^{-7} \text{ cm}^3/\text{sec.} \times (760/760) \times (273/299) = 9.95 \times 10^{-8} \text{ cm}^3/\text{sec}$$

3) The permeability constant, p, is calculated using the following equation;

$$p = q \text{ (STP)} \times L / (A \times DP)$$

**where:**

L = test specimen thickness, em,

A = cross-sectional area of test specimen, cm<sup>2</sup>, and

DP = pressure differential across test specimen, atm.

For example:

$$p = \frac{(9.95 \times 10^{-8} \text{ cm}^3/\text{sec}) (0.20 \text{ em})}{(66.4 \text{ cm}^2) (3.40 \text{ atm})}$$

$$= 8.81 \times 10^{-11} \text{ cm}^2/\text{sec-atm}$$



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Presented this 26th day of May 2010.



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